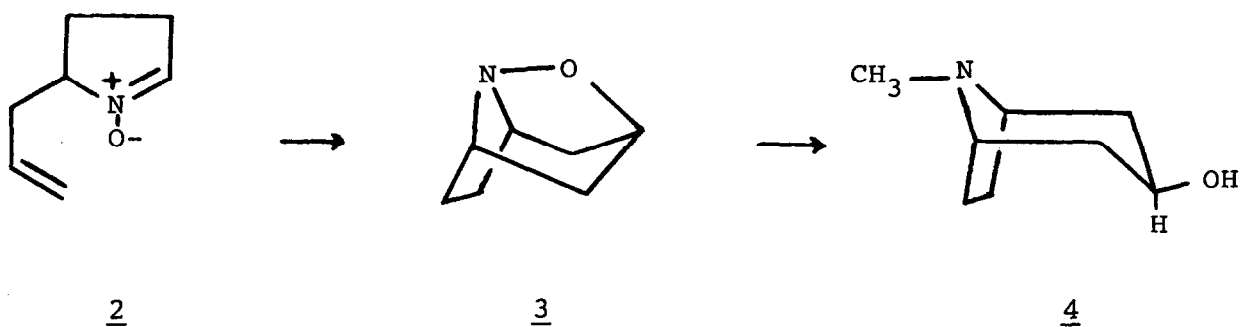


A STEREOSPECIFIC SYNTHESIS OF (+)-COCAINE

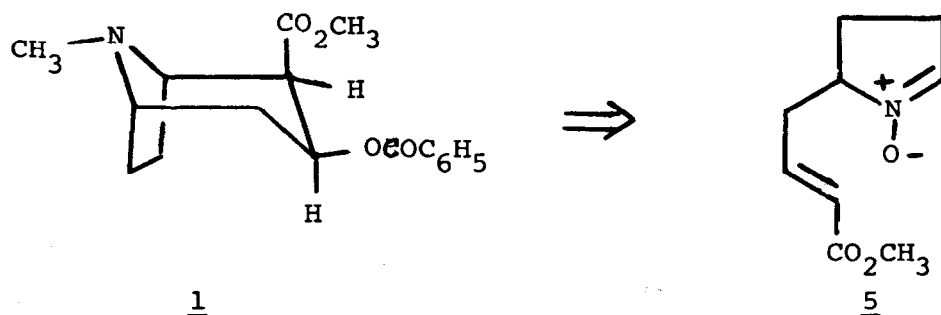
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Efficient nitron-based entries into the tropane skeleton have been described^{1,2}. Thus, nitron olefin 2 was efficiently converted into pseudotropine 4, via cycloadduct 3, according to the following scheme².

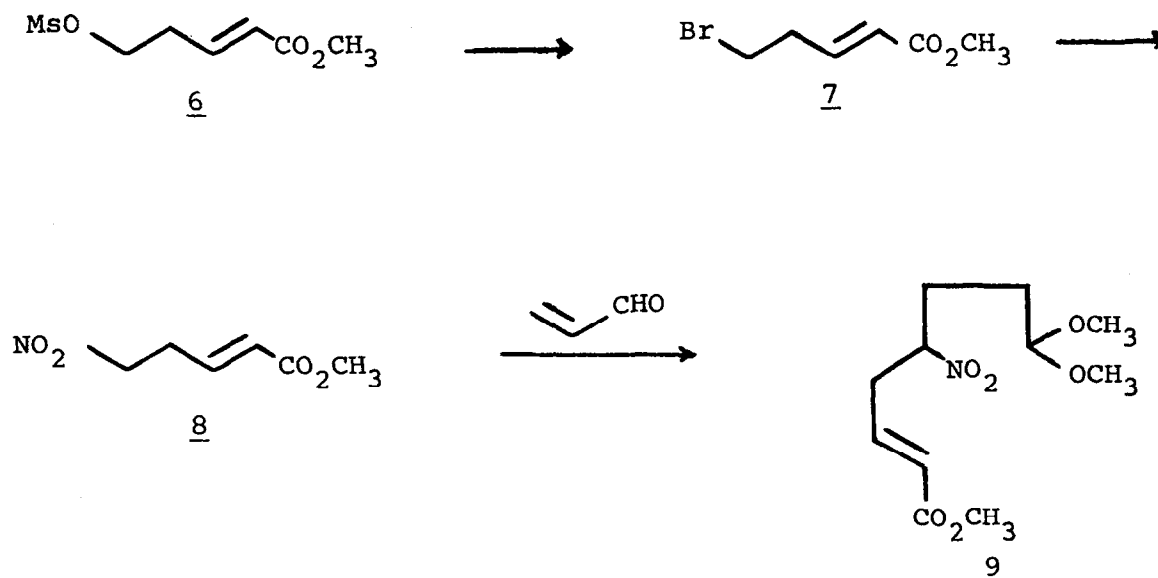


We wish to report herein an approach to cocaine (1) which emphasizes the high degree of regiochemical and stereochemical control possible using nitron cycloadditions. Prior synthetic efforts toward cocaine have confronted severe difficulties relating to the relative stereochemistry of the two ester functions³⁻⁶. Our stereospecific approach depends on the synthesis of nitron ester 5 as the key precursor of the natural product. Our initial



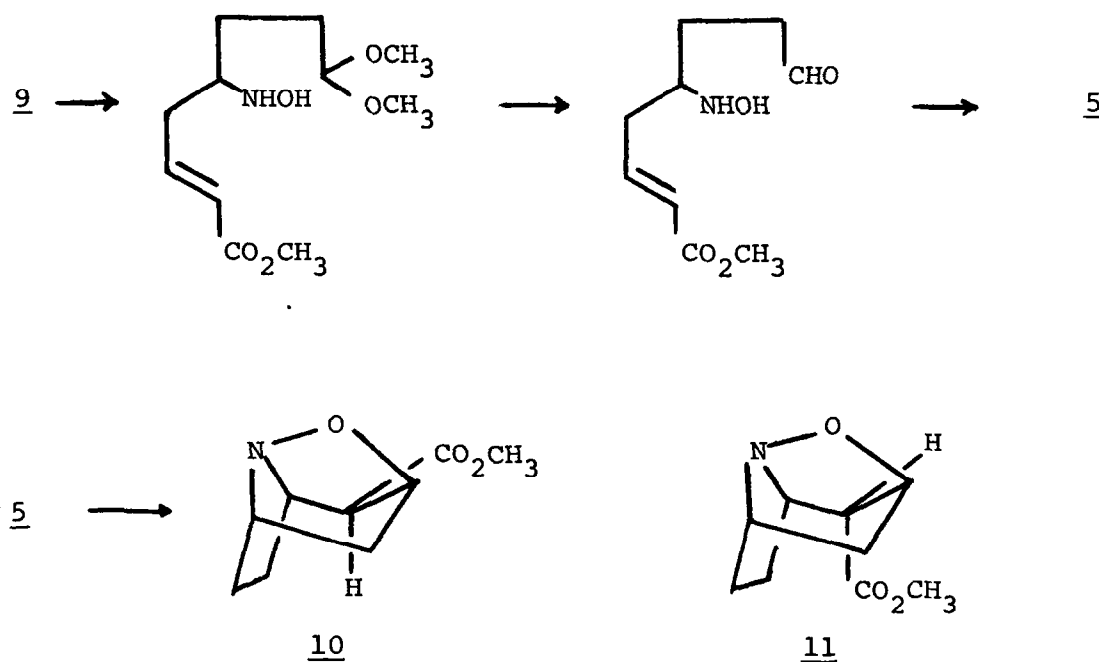
objective is therefore the synthesis of this nitron ester (5). To accomplish this, we converted mesylate olefin 6⁷ into the corresponding bromo ester 7 (80%) using lithium bromide in ether. The nitro compound 8 was then prepared from 7 using silver nitrite in

acetonitrile. Michael addition of nitro olefin 8 onto acrolein using sodium methoxide in methanol, followed in situ protection of the aldehyde functionality, provided acetal 9 in 90% overall yield.



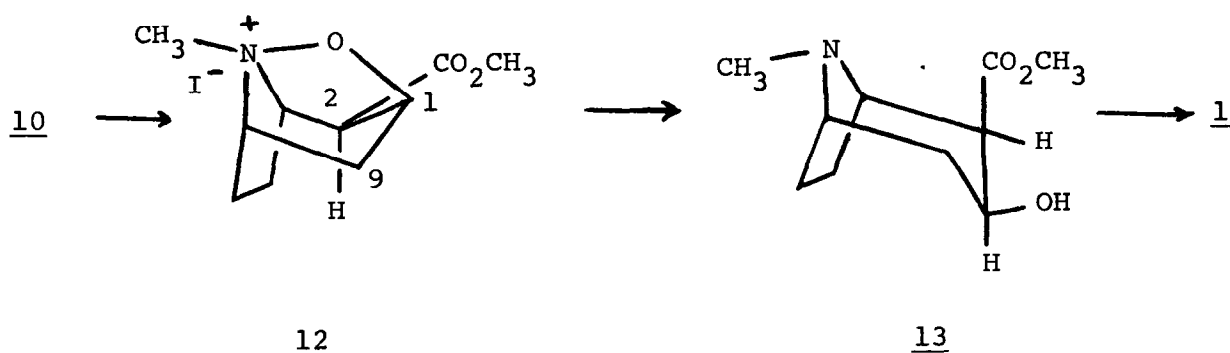
The conversion of 9 into the desired nitron ester 5 was accomplished by reduction of the nitro group with zinc and ammonium chloride in dimethoxyethane-water solution. After acidification, the resulting oil was heated in toluene to afford ca. 10% of a white solid, identified as tricyclic ester 10⁸. The probable sequence of reactions involved is indicated below. The nitron 5 was not characterized but, rather, directly cyclized to 10.

The ir spectrum of 10 exhibits a band at 5.80 μ , suggesting the retention of the ester functionality. The nmr spectrum exhibits a one-proton doublet at δ 4.95 ppm (J 5.5 Hz)



assignable to the C-1 proton since it is coupled only to the exo-proton at C-9. A dihedral

angle of 90° exists between the C-1 proton and the endo-protons at C-2 and C-9. This observation is consistent with the nmr spectrum of 3, the intermediate in the synthesis of pseudotropine⁹. One proton multiplets at δ 3.53 and 3.86 ppm were assignable to the two methine protons at C-3 and C-6. Methylation of the cycloadduct 10 with excess methyl iodide in methylene chloride and ether at reflux for 24 hours provided the methiodide 12.



The ir spectrum of 12 contained the ester carbonyl stretch at 5.79μ . The nmr spectrum contained two three-proton singlets at δ 3.70 and 3.80 ppm for the quaternary ammonium methyl group and the methyl ester, respectively, and two one-proton multiplets at δ 4.42 and 4.76 ppm for the methine protons at C-3 and C-6, along with a characteristic one-proton doublet ($J = 5.5\text{Hz}$) at δ 5.58 ppm for the C-1 proton. A primary advantage of the nitron-based approach is the ability to control the stereochemistry of the ester function. Thus, as expected, no adduct (i.e. 11) with an endo-carbomethoxyl group was observed. The (Z)-configuration at the olefinic center in 5 fixes the stereochemistry of the carbomethoxyl group in 10 and 12. This in turn provides for the axial ester grouping in ($^+$)-cocaine (vide infra).

The nitrogen oxygen bond of methiodide adduct 12 was cleaved with zinc and acetic acid to give a 47% yield of ecgonine methyl ester (13). The ir spectrum exhibited strong absorption at 2.60μ confirming the presence of a free hydroxyl group, and a strong band at 5.86μ , attributable to the ester carbonyl stretch. The nmr spectrum displayed the expected three-proton singlet at δ 1.90 ppm (CH_3N) and another at δ 3.28 ppm (CO_2CH_3).

Benzoylation of hydroxyester 13 using the method of de Jong⁴ provided ($^+$)-cocaine whose ir, nmr, and mass spectra were virtually identical with those of natural ($-$)-cocaine. Thus, the intramolecular nitron cyclization has provided the first stereospecific total synthesis of ($^+$)-cocaine.

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